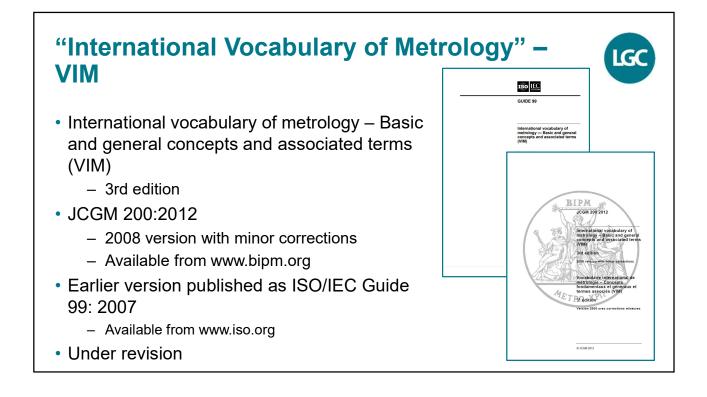






- Basic principles of measurement are the same across disciplines
 - Physics, chemistry, laboratory medicine, biology, engineering...
 - Increase in interdisciplinary activities
- Clear definitions
 - Improve communication
 - Reduce disputes/misunderstanding
 - Allow consistent interpretation
 - E.g. between regulatory/assessment bodies and laboratories
 - Facilitate translations

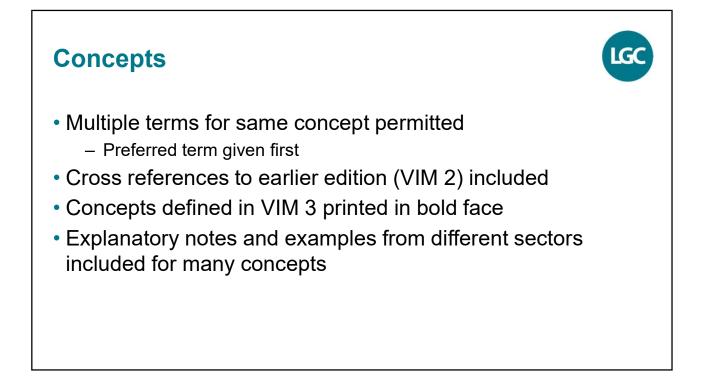
You talk, we und	
The way out of the t	ower of Babel
he problem	
te live in a "global village" but we are all different and we se many different languages to communicate.	Recurso
on when pools pools the same language, the same language, the same language of the same language strange set of the same language strange set of the same set of additional strange set of additional strange set of the same set of additional strange set of additional s	nft. The may lead to inadequate furthered
common language	
e need a common language; clear and unambiguous. V insistent definitions of concepts with their associated terms.	le need a common vocabulary to provide
b this end, several organisations participate in a joint frot to develop and maintain an international actuality of metrodogy (VBN) [1], which aims to hieve a unique understanding of concepts related to hieve a unique understanding of concepts related to accurate the several sectors. We VIM is a normative reference in the standards output JUNS, 150 15189 and 150/EEC 17043, and is the exempt accurations.	
s this enough?	
In laboratory staff some problems still stand and need to be Many people are confused about toth concepts and terms It may be difficult to understand the "formal" VM defit statements intended to apply to many different measures Translators into local languages may add to the con- terms are used in different sectors for the same VIM con- overlook the issue.	s, sitions – they are short nent sectors. fusion, e.g. if different icept and the translator
	Eurachem
	A FOCUS FOR ANALYTICAL CHEMISTRY IN ELMORE



VIM 3 Content



- Concepts listed in five chapters
 - Chapter 1 Quantities and Units
 - Chapter 2 Measurement
 - Chapter 3 Devices for measurement
 - Chapter 4 Properties of measuring devices
 - Chapter 5 Measurement standards (Etalons)
- 12 Concept diagrams



Concept - example



5.1 (6.1)

measurement standard

Etalon

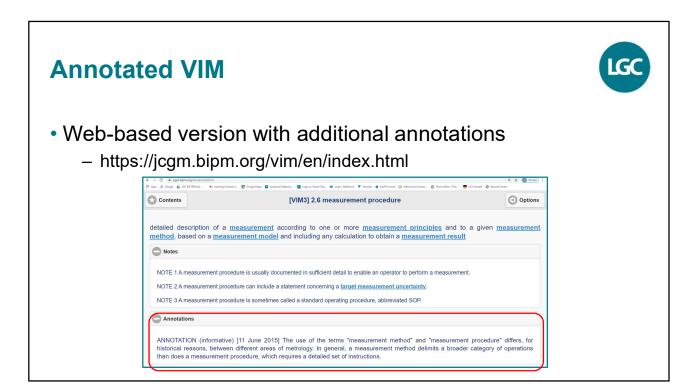
realization of the definition of a given **quantity**, with stated **quantity value** and associated **measurement uncertainty**, used as a reference EXAMPLE 1 1 kg mass measurement standard with an associated standard measurement uncertainty of 3 µg.

EXAMPLE 6 **Reference material** providing quantity values with measurement uncertainties for the mass concentration of each of ten different proteins.

NOTE 1 A "realization of the definition of a given quantity" can be provided by a measuring system, a material measure, or a reference material.



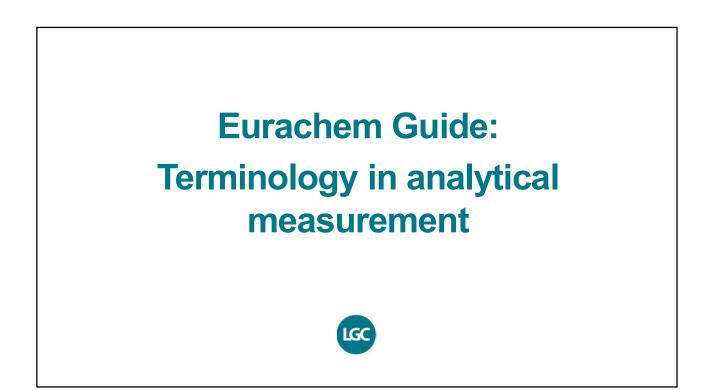
. . .



Development of "VIM 4"

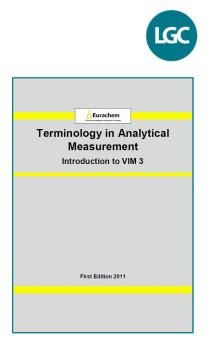


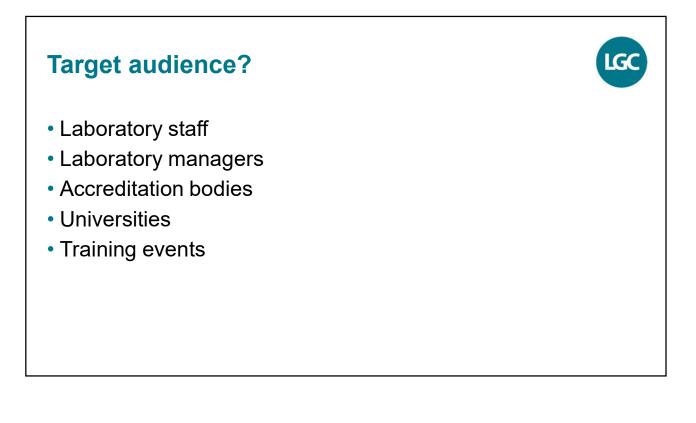
- VIM is under revision
 - Joint Committee for Guides in Metrology (JCGM) WG 2
- Incorporate annotations from the web-based version
- Include essential definitions for nominal properties
- Simplify language



Scope of the Eurachem Guide Applicable to several sectors Chemical analysis Biological measurements Clinical chemistry Important concepts identified Terms requiring special attention identified Link to terms used in ISO standards and

Guides – Current terminology in the workplace

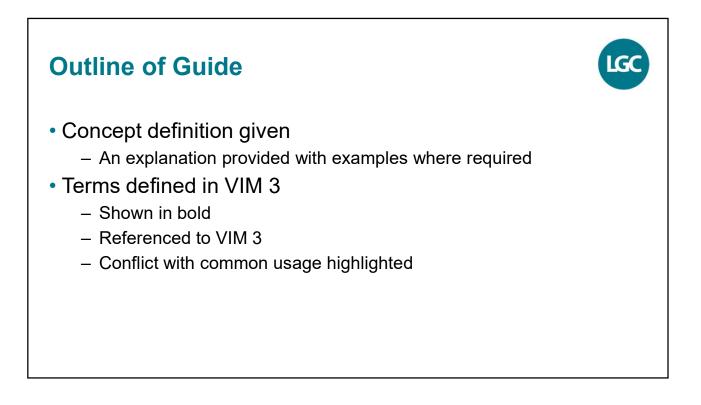




Layout of guide



- Terms divided into 'families'
 - General metrology
 - Metrological traceability
 - Measurement uncertainty
 - Validation/verification and method performance
- Potential problem terms identified within family
- Terms frequently used in routine laboratories



Potential problem terms



- Measurement procedure
- Calibration (curve and diagram)
- Validation and verification
- Measuring interval

